

- 1 A heat engine absorbs 250 J of heat from a hot reservoir and rejects 110 J to a cold reservoir. What is the efficiency of this engine?
- A) 39.% B) 78.% C) 31.% D) 85.% E) 56.%
- 2 A heat engine with an output of 300 W has an efficiency of 30% and works at a frequency of 10 cycles/second. How much heat is absorbed (Q_h) and how much rejected (Q_c) in each cycle?
- A) $Q_h = 200. \text{ J}$ $Q_c = 140. \text{ J}$ B) $Q_h = 143. \text{ J}$ $Q_c = 9.00 \text{ J}$ C) $Q_h = 1000. \text{ J}$ $Q_c = 700. \text{ J}$
 D) $Q_h = 100. \text{ J}$ $Q_c = 70.0 \text{ J}$ E) $Q_h = 129. \text{ J}$ $Q_c = 90.0 \text{ J}$

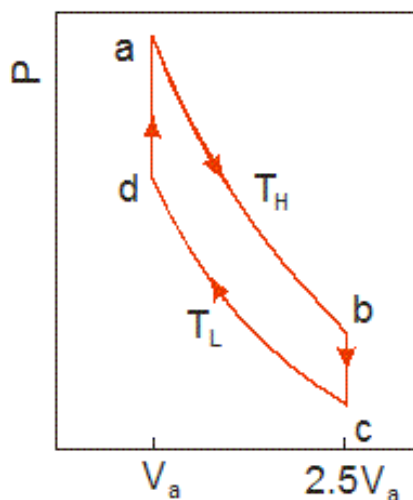


Figure 1: Problem 3

- 3 Refer to Fig. 1. An ideal heat engine uses 0.300 mol of ideal gas and operates between a hot reservoir at $T_H = 400 \text{ K}$ and cold reservoir at $T_L = 300 \text{ K}$, in a cycle from $a \rightarrow b \rightarrow c \rightarrow d \rightarrow a$. From $a \rightarrow b$ the gas undergoes an isothermal expansion, changing its volume from V_a to $2.5V_a$. From $b \rightarrow c$, the pressure is reduced at a constant volume. From $c \rightarrow d$, the gas undergoes an isothermal compression, and from $d \rightarrow a$, the pressure is increased at a constant volume until the gas is back at the original condition at a. How much heat is absorbed in going from $a \rightarrow b$?
- A) 1100. J B) 686. J C) 823. J D) 3050. J E) 914. J
- 4 Two refrigerators, one with a COP of 3.5 and another with a COP of 5.0, both extract 400 kJ of heat from the cold reservoir (food). Calculate the difference in energy they exhaust to the hot reservoir and hence the room.
- A) 34. kJ B) 24. kJ C) 38. kJ D) 31. kJ E) 41. kJ
- 5 A water-cooled electric power plant generates 250 MW of power at an efficiency of 35.0%. At what rate must water be circulated past the condenser if the change in water temperature is not to exceed 10° C ? (The specific heat of water is $4.2 \times 10^3 \text{ J/kg}^\circ \text{ C}$)
- A) 11100. kg/s B) 8840. kg/s C) 2630. kg/s D) 6630. kg/s E) 111000. kg/s
- 6 What is the maximum possible coefficient of performance of a heat pump that is capable of maintaining the interior of a house at 28° C when the temperature outside is -40° C ?
- A) 0.44 B) 0.41 C) 4.4 D) 0.56 E) 3.4